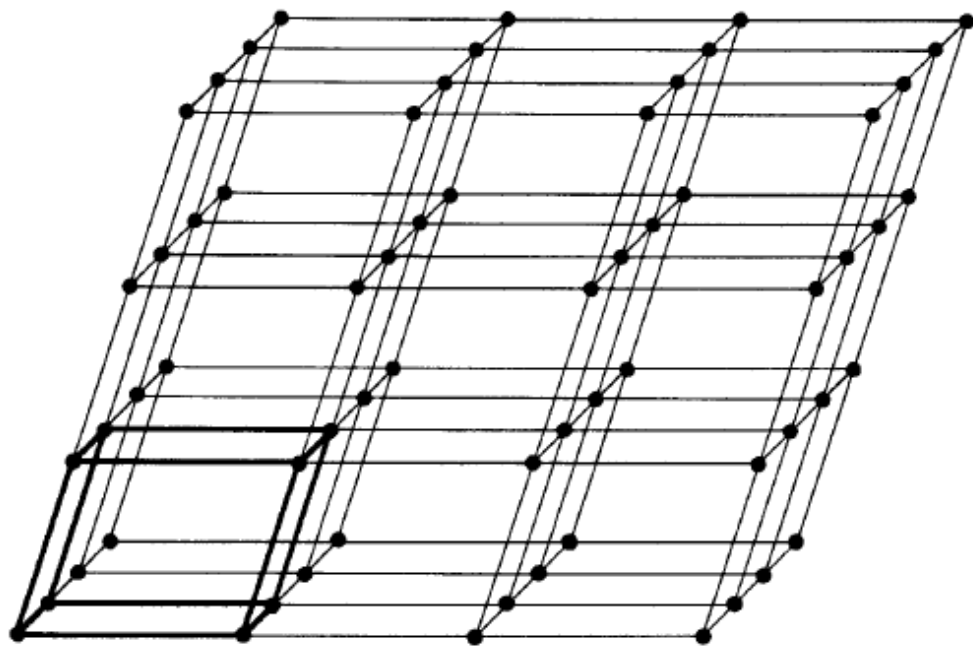




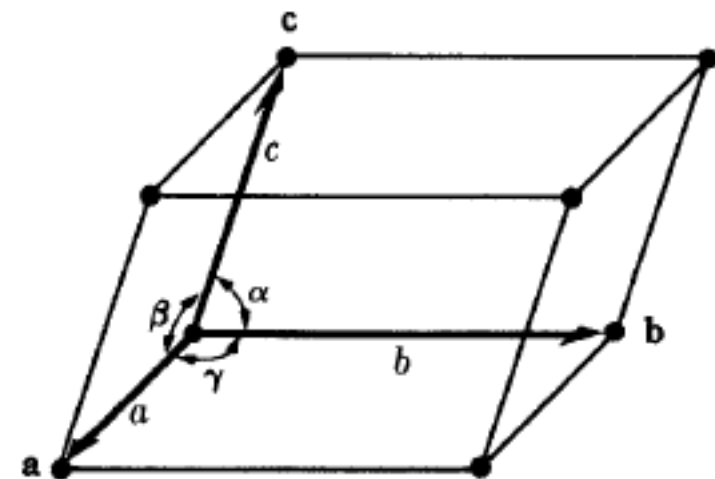
# Crystallography

Mohammad Eghbali

# شبکه و یکایاخته

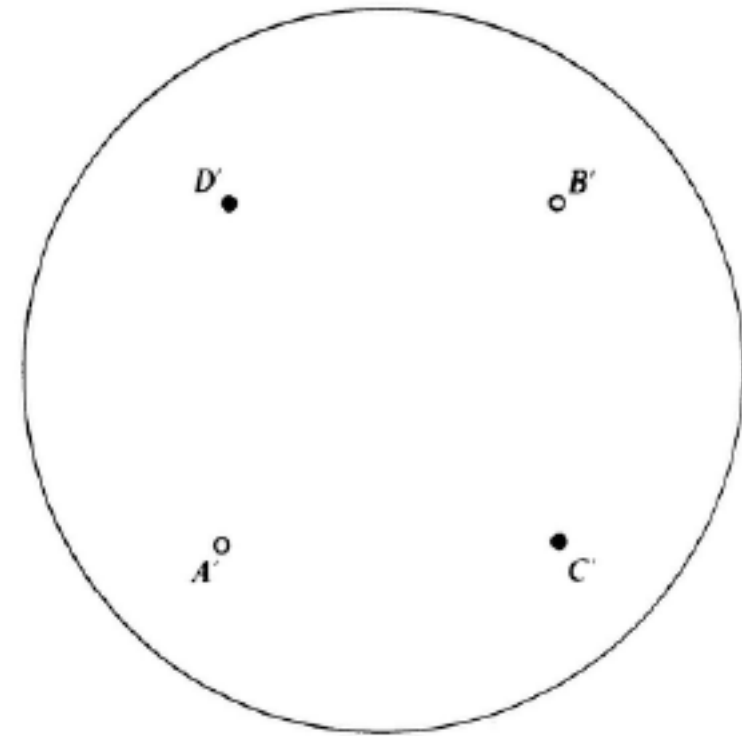
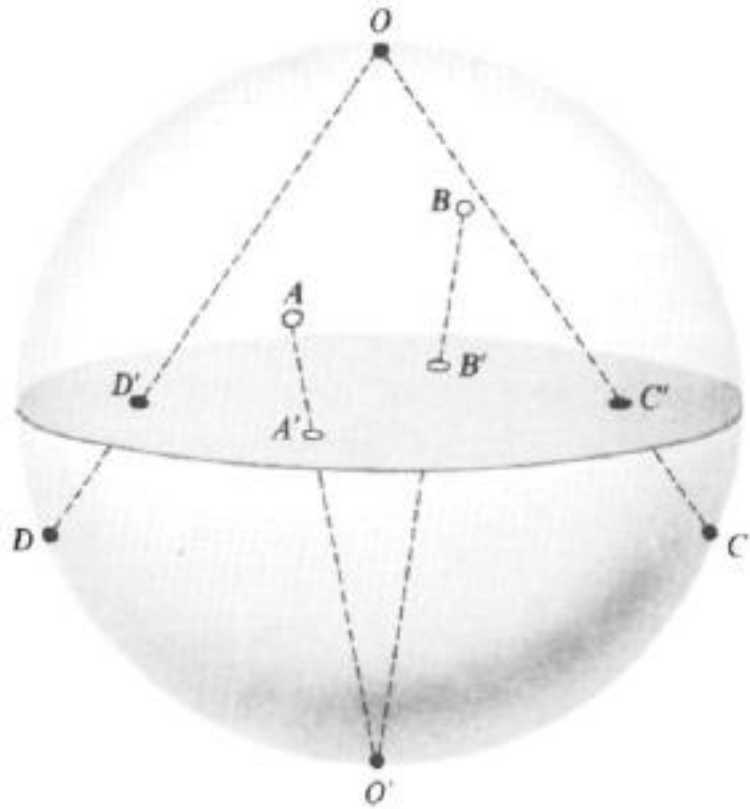


شبکه نقطه ای



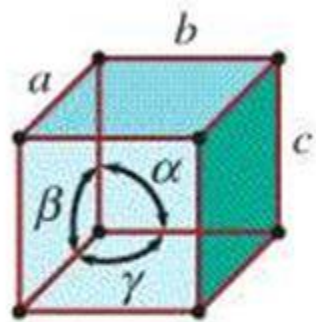
یکا یاخته

# symmetry of crystals

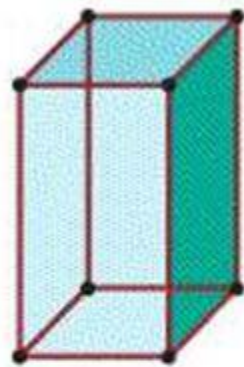


The stereographic projection

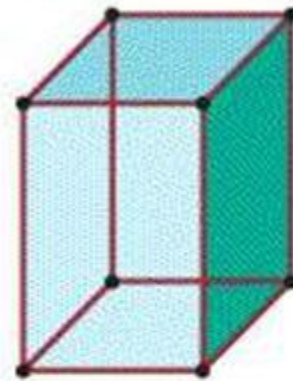
# Seven Types of Unit Cells



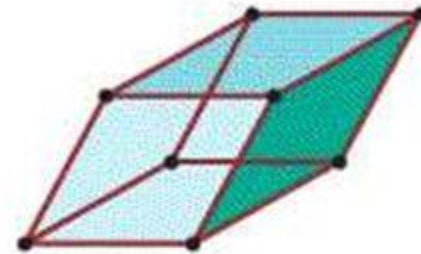
**Simple cubic**  
 $a = b = c$   
 $\alpha = \beta = \gamma = 90^\circ$



**Tetragonal**  
 $a = b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$



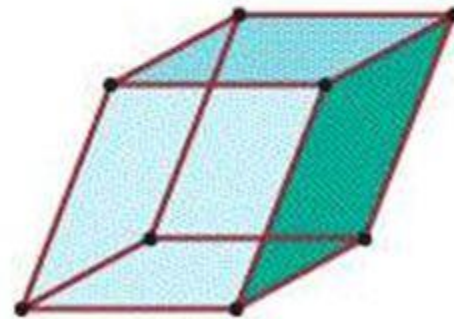
**Orthorhombic**  
 $a \neq b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$



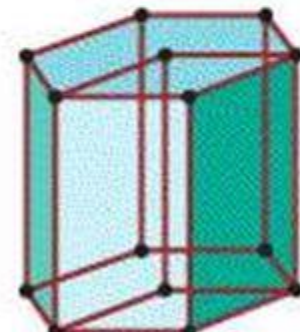
**Rhombohedral**  
 $a = b = c$   
 $\alpha = \beta = \gamma \neq 90^\circ$



**Monoclinic**  
 $a \neq b \neq c$   
 $\alpha = \gamma = 90^\circ, \beta \neq 90^\circ$

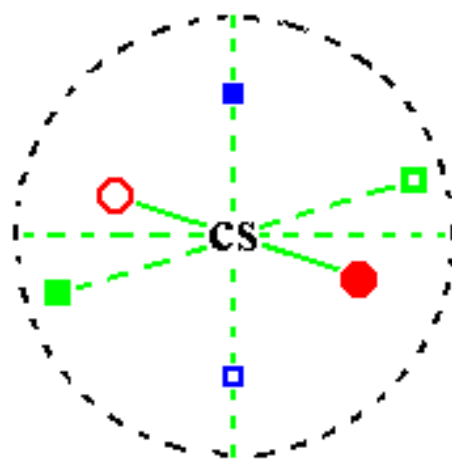
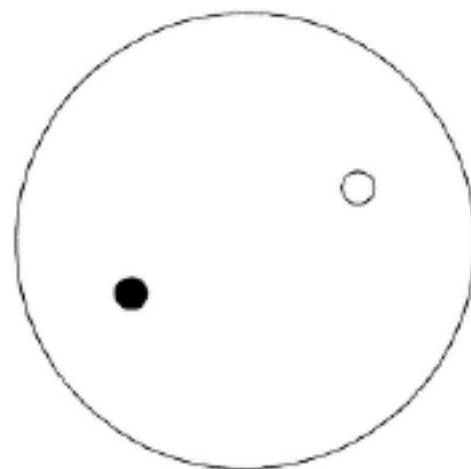


**Triclinic**  
 $a \neq b \neq c$   
 $\alpha \neq \beta \neq \gamma \neq 90^\circ$

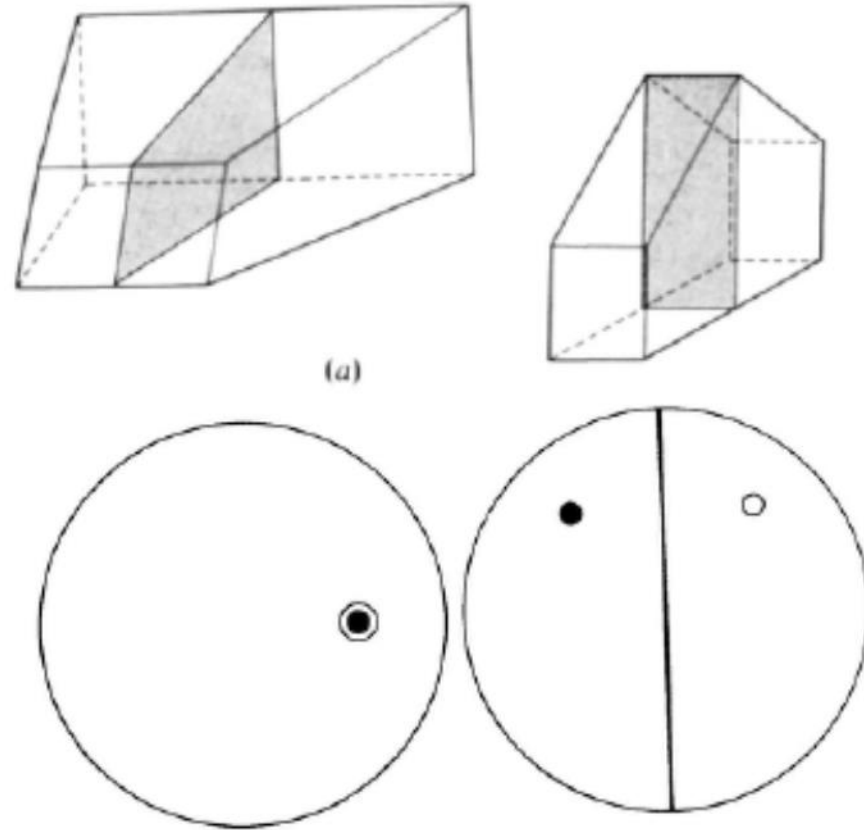



**Hexagonal**  
 $a = b \neq c$   
 $\alpha = \beta = 90^\circ, \gamma = 120^\circ$

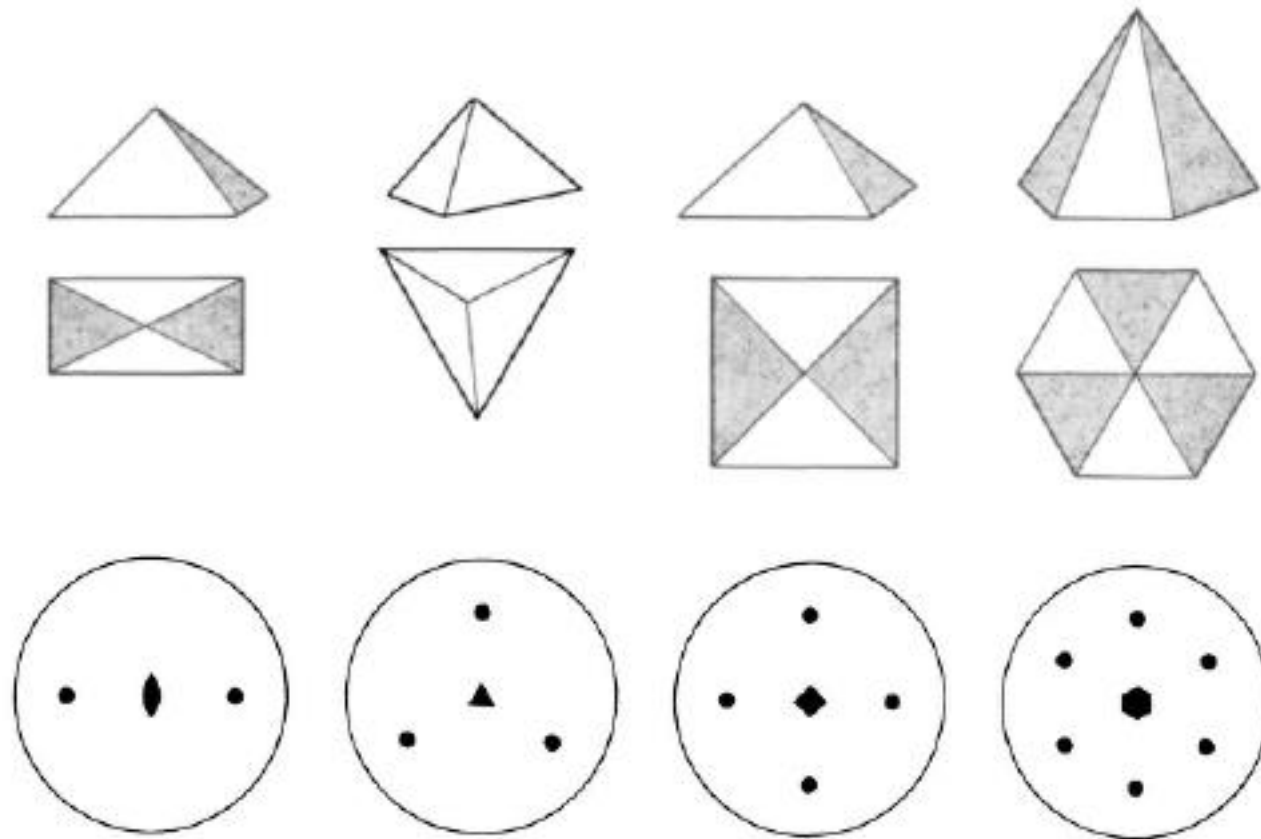
Centre of symmetry



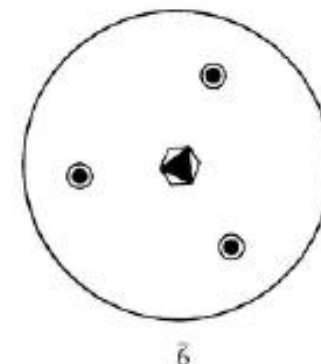
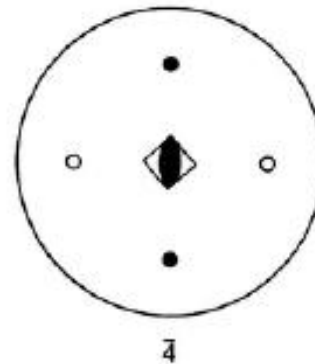
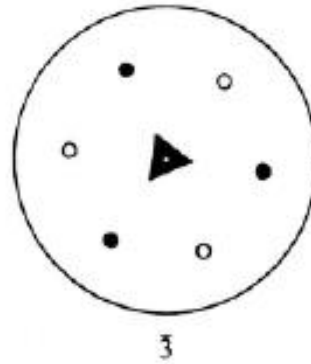
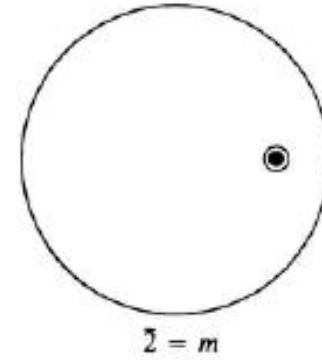
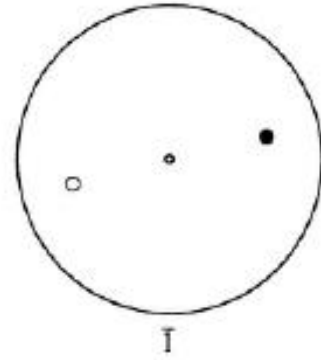
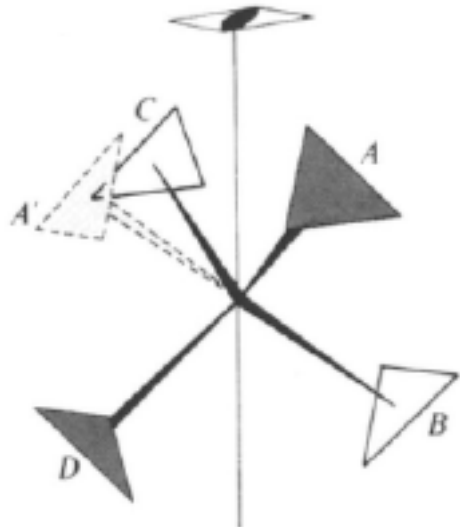
Mirror plane (written symbol  $m$ ; graphical symbol  $\text{—}$ )



Rotation axes (written symbols 2, 3, 4, 6; graphical symbols )



Inversion axes (written symbols 1, 2, 3, 4, 6; graphical symbols  $\bar{1}$ ,  $\bar{2}$ ,  $\bar{3}$ ,  $\bar{4}$ ,  $\bar{6}$ )



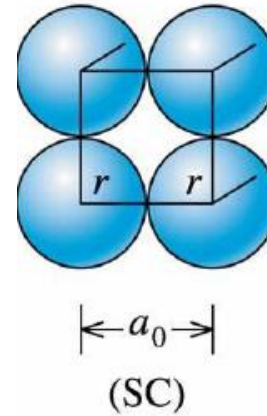
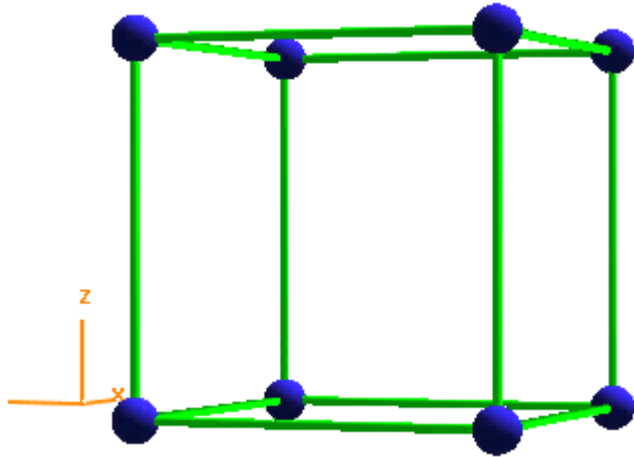


# Bravais lattice

*simple cubic (sc)*

$$a=b=c$$

$$\alpha=\beta=\gamma=90$$



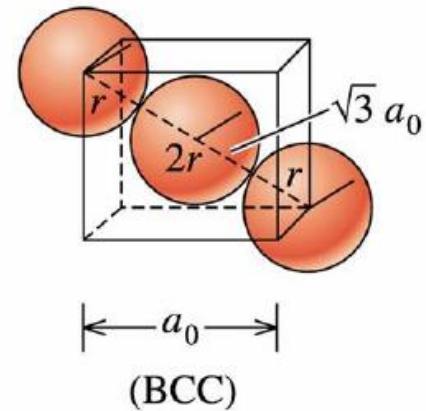
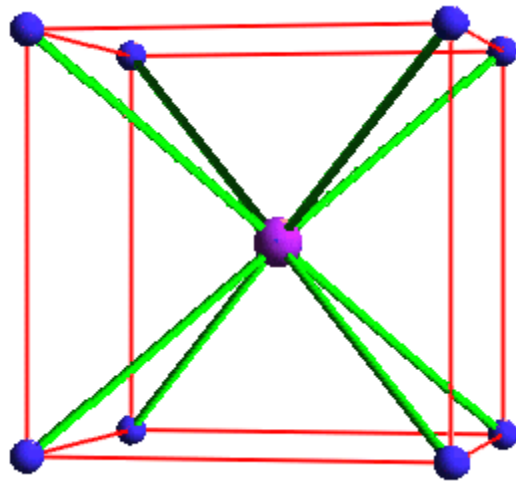
$$APF = \frac{\frac{4}{3} \pi \left(\frac{a}{2}\right)^3}{a^3} = \frac{\pi}{6}$$

# Bravais lattice

*body centered cubic (bcc)*

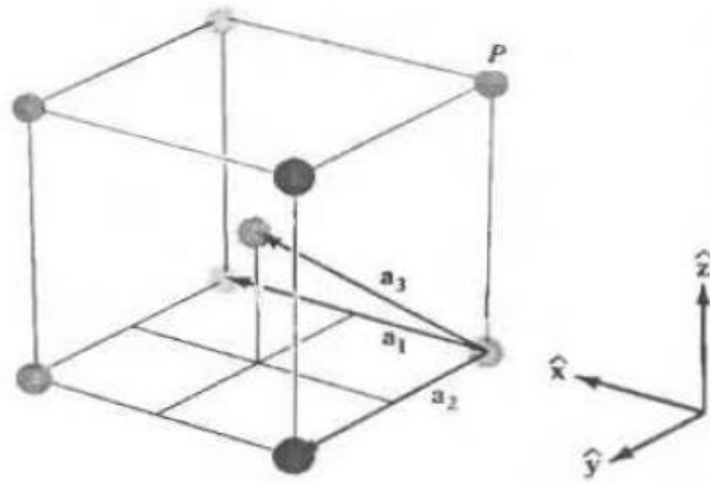
$$a=b=c$$

$$\alpha=\beta=\gamma=90$$

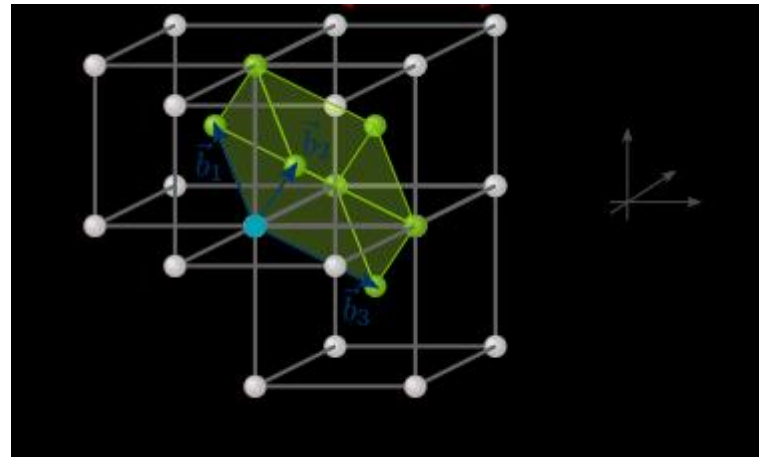
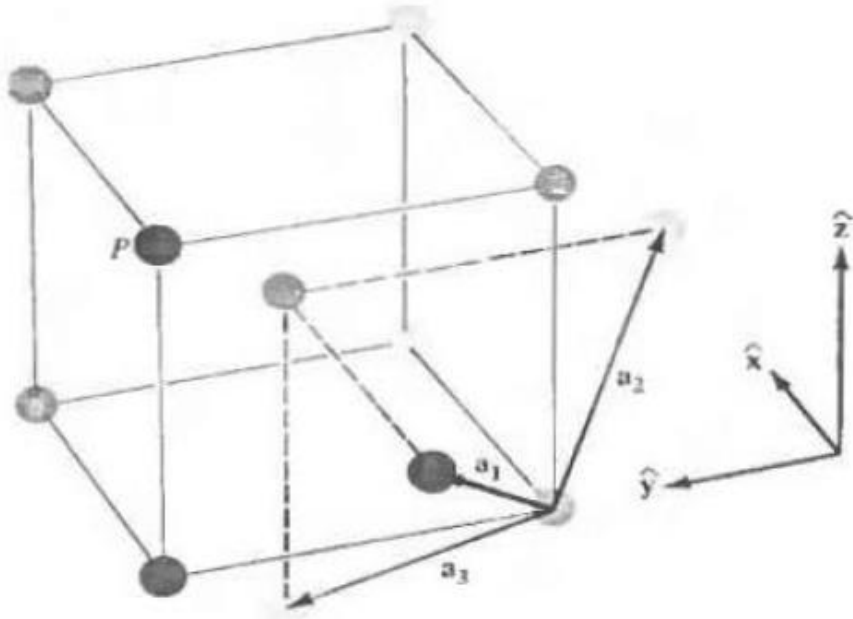
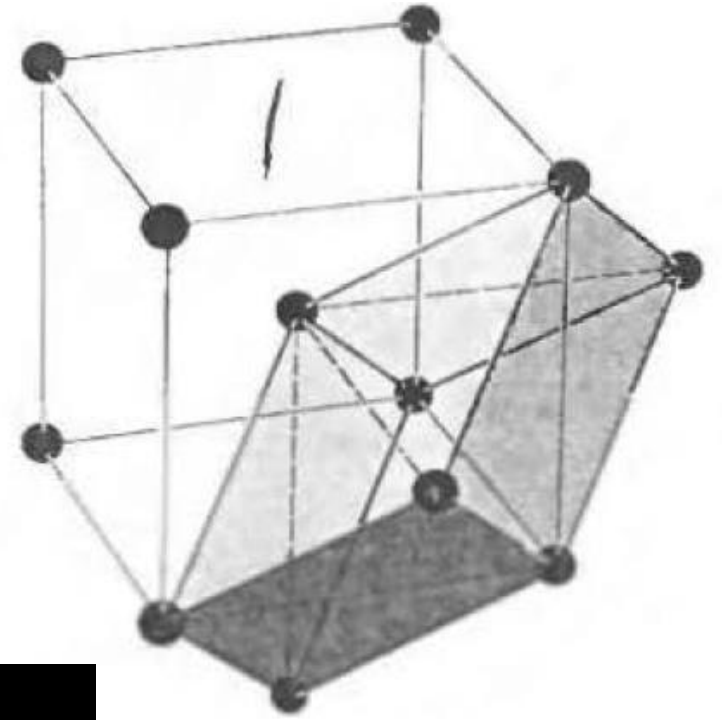


$$APF = \frac{2 \times \frac{4}{3} \pi \left( \frac{\sqrt{3}}{4} a \right)^3}{a^3} = \frac{\sqrt{3}}{8} \pi$$

bcc



$$\mathbf{a}_1 = a\hat{x}, \quad \mathbf{a}_2 = a\hat{y}, \quad \mathbf{a}_3 = \frac{a}{2}(\hat{x} + \hat{y} + \hat{z}).$$



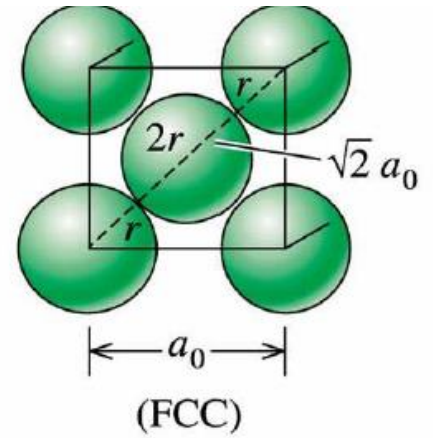
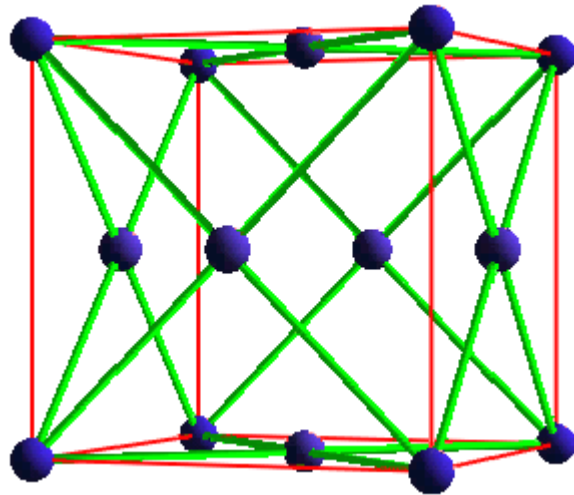
$$\mathbf{a}_1 = \frac{a}{2}(\hat{y} + \hat{z} - \hat{x}), \quad \mathbf{a}_2 = \frac{a}{2}(\hat{z} + \hat{x} - \hat{y}), \quad \mathbf{a}_3 = \frac{a}{2}(\hat{x} + \hat{y} - \hat{z}).$$

# Bravais lattice

*face centered cubic (fcc)*

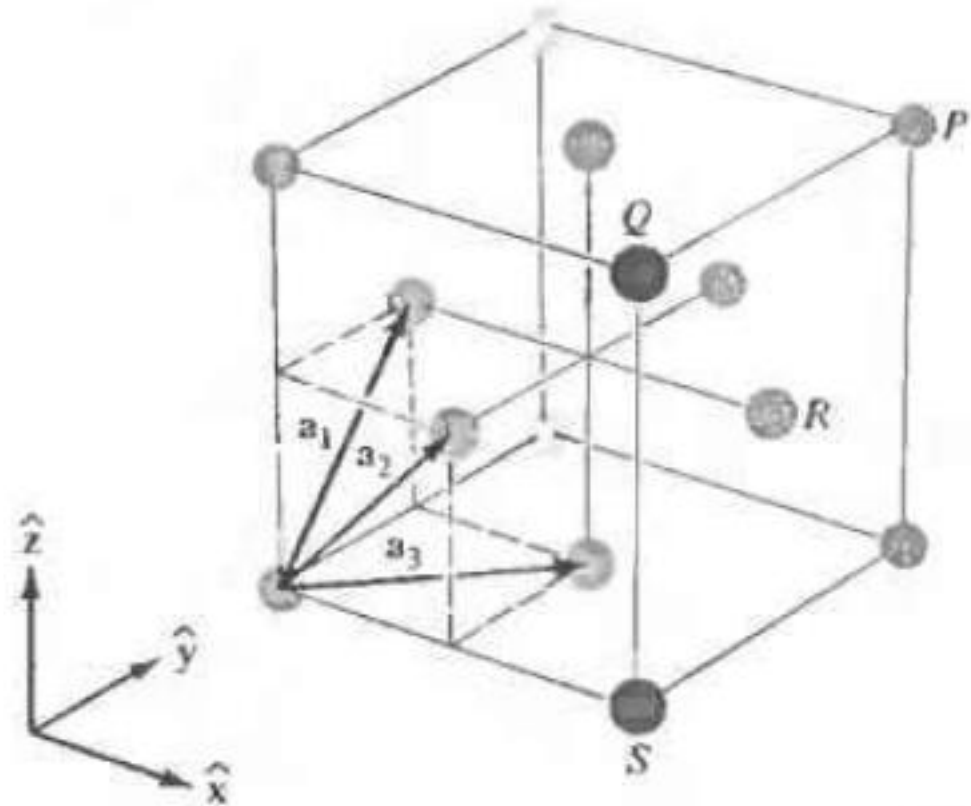
$$a=b=c$$

$$\alpha=\beta=\gamma=90$$

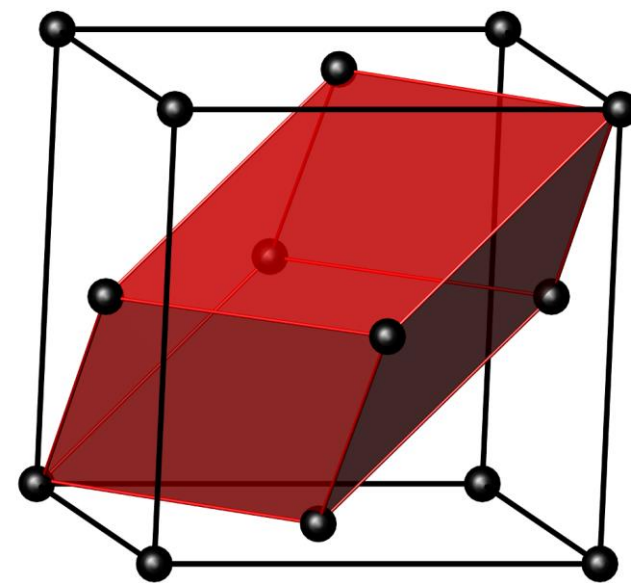
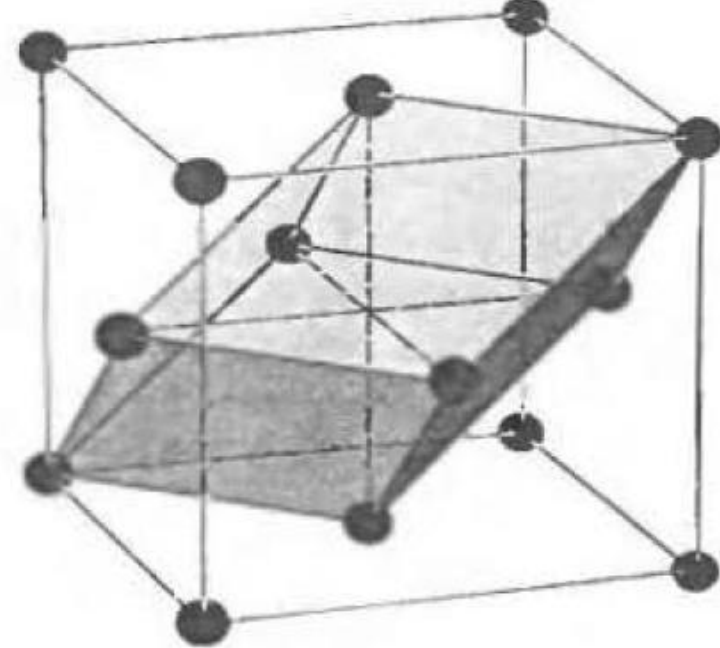


$$APF = \frac{4 \times \frac{4}{3} \pi \left( \frac{\sqrt{2}}{4} a \right)^3}{a^3} = \frac{\sqrt{2}}{6} \pi$$

fcc

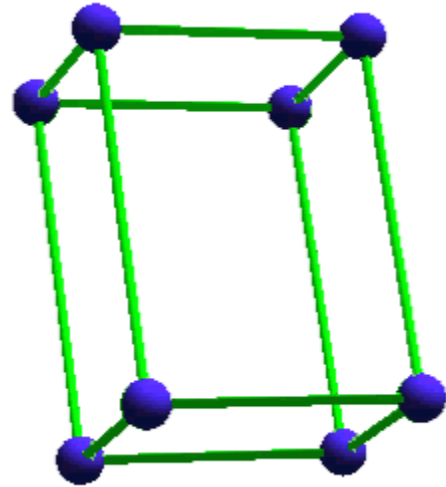


$$\mathbf{a}_1 = \frac{a}{2}(\hat{y} + \hat{z}), \quad \mathbf{a}_2 = \frac{a}{2}(\hat{z} + \hat{x}), \quad \mathbf{a}_3 = \frac{a}{2}(\hat{x} + \hat{y}).$$



# *Bravais lattice*

*orthohombic (p)     $a \neq b \neq c$      $\alpha = \beta = \gamma = 90$*



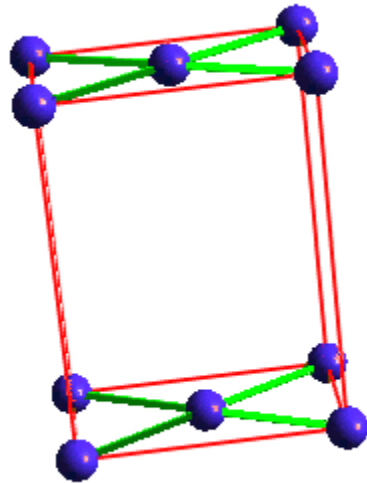
$$APF = \frac{\frac{4}{3} \pi \left( \frac{a}{2} \right)^3}{abc}$$

# *Bravais lattice*

*orthorhombic (c)*

$$a \neq b \neq c$$

$$\alpha = \beta = \gamma = 90$$



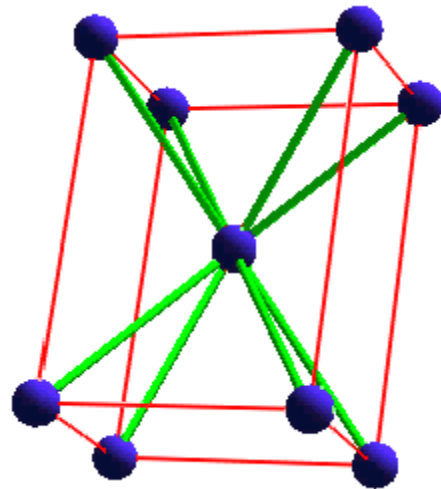
$$APF = \frac{2 \times \frac{4}{3} \pi \left( \frac{\sqrt{a^2 + b^2}}{4} \right)^3}{abc}$$

# *Bravais lattice*

*orthohombic (I)*

$a \neq b \neq c$

$\alpha = \beta = \gamma = 90$



$$APF = \frac{2 \times \frac{4}{3} \pi \left( \frac{\sqrt{a^2 + b^2 + c^2}}{4} \right)^3}{abc}$$

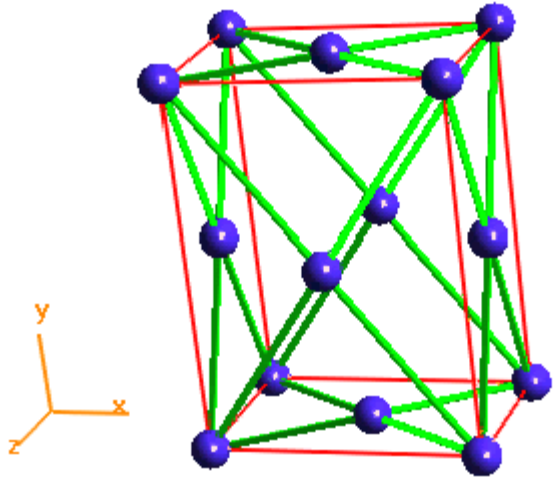


# *Bravais lattice*

*orthohombic (f)*

$a \neq b \neq c$

$\alpha = \beta = \gamma = 90$

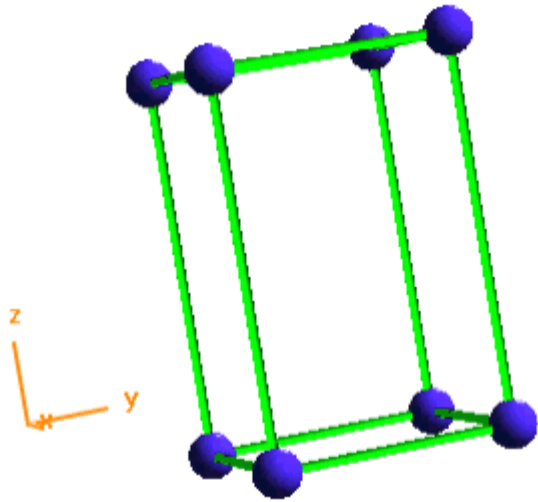


$$APF = \frac{4 \times \frac{4}{3} \pi \left( \frac{\sqrt{a^2 + b^2}}{4} \right)^3}{abc}$$

# Bravais lattice

tetragonal (p)

$$a=b \neq c \quad \alpha=\beta=\gamma=90^\circ$$

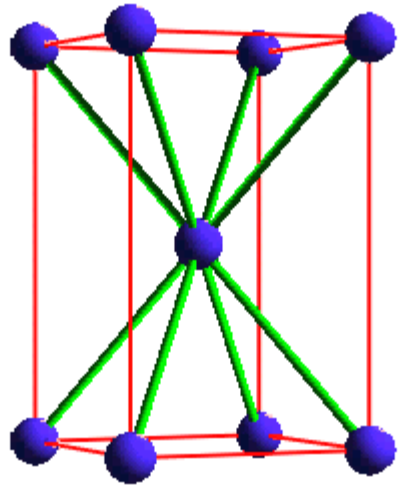


$$APF = \frac{\frac{4}{3} \pi \left( \frac{a}{2} \right)^3}{a^2 c} = \frac{\pi a}{6 c}$$

# *Bravais lattice*

*tetragonal (I)*

$$a=b \neq c \quad \alpha=\beta=\gamma=90$$



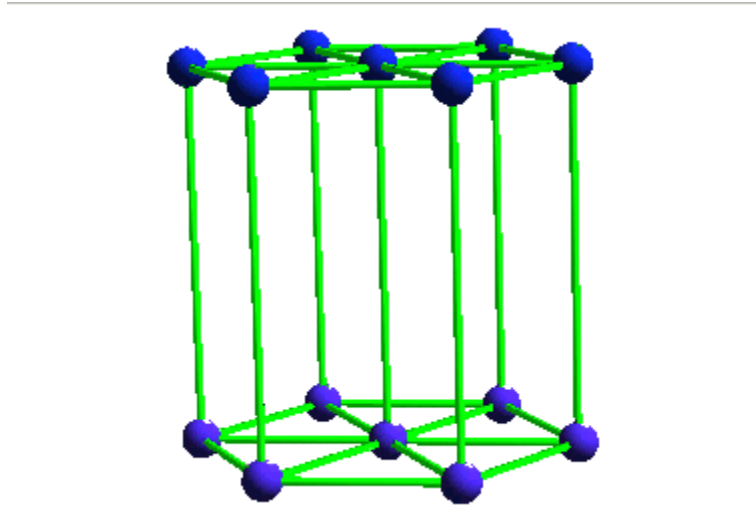
$$APF = \frac{2 \times \frac{4}{3} \pi \left( \frac{a}{2} \right)^3}{a^2 c}$$

$$APF = \frac{2 \times \frac{4}{3} \pi \left( \frac{\sqrt{2a^2 + c^2}}{4} \right)^3}{a^2 c}$$

# *Bravais lattice*

*hexagonal*

$$a=b \neq c \quad \alpha=\beta=90^\circ, \gamma=120^\circ$$



$$APF = \frac{3 \times \frac{4}{3} \pi \left( \frac{a}{2} \right)^3}{3 \times \frac{\sqrt{3}}{2} a^2 c}$$

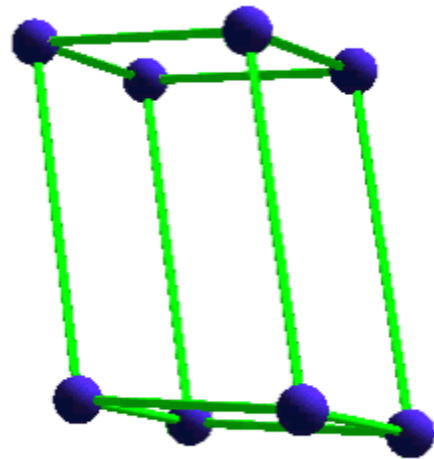
# *Bravais lattice*

*monoclinic*

$a \neq b \neq c$

$\alpha = \gamma = 90$

$\beta$  is not equal 90



$$APF = \frac{\frac{4}{3} \pi \left(\frac{a}{2}\right)^3}{abc \sin \beta}$$

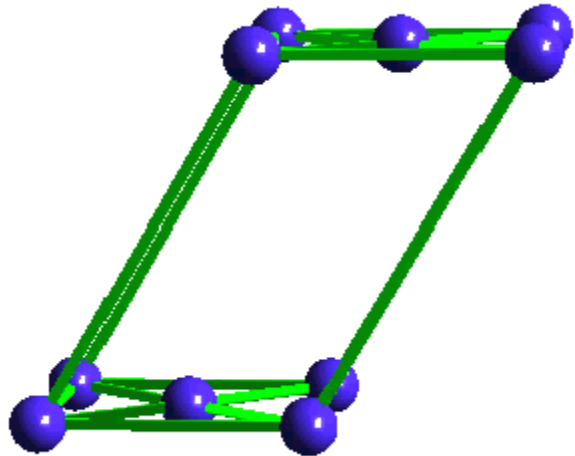
# *Bravais lattice*

*monoclinic*

$a \neq b \neq c$

$\alpha = \gamma = 90$

$\beta$  is not equal 90



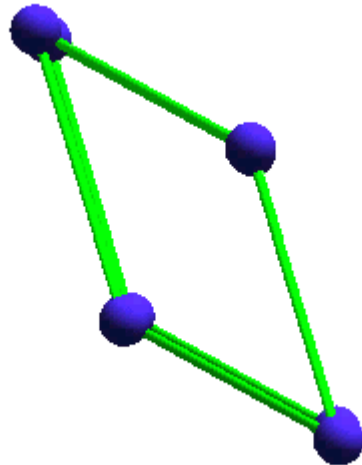
$$APF = \frac{2 \times \frac{4}{3} \pi \left(\frac{a}{2}\right)^3}{abc \sin \beta}$$

# *Bravais lattice*

*triclinic*

$a \neq b \neq c$

$\alpha \neq \beta \neq \gamma \neq 90$

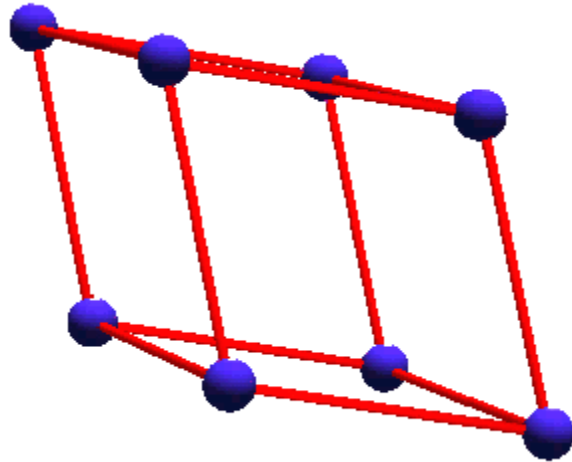


$$AFP = \frac{\frac{4}{3} \pi \left(\frac{a}{2}\right)^3}{abc \sqrt{1 - \cos^2 \alpha - \cos^2 \beta - \cos^2 \gamma + 2 \cos \alpha \cos \beta \cos \gamma}}$$

# *Bravais lattice*

*trigonal(rhombohedral)*

$$a=b=c \quad \alpha=\beta=\gamma \neq 90$$



$$APF = \frac{\frac{4}{3} \pi \left(\frac{a}{2}\right)^3}{a^3 \sqrt{1 - 3 \cos^2 \alpha + 2 \cos^3 \alpha}}$$



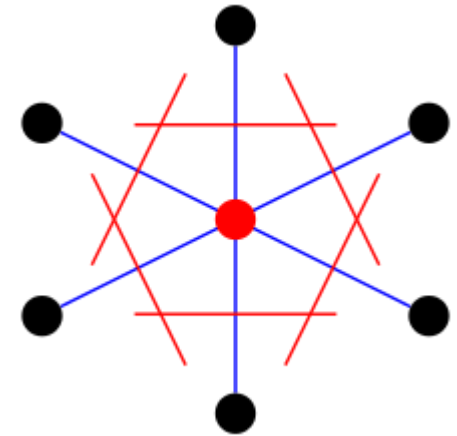
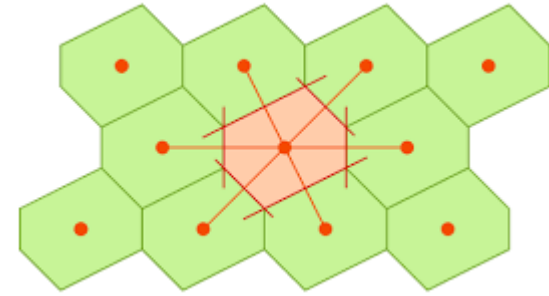
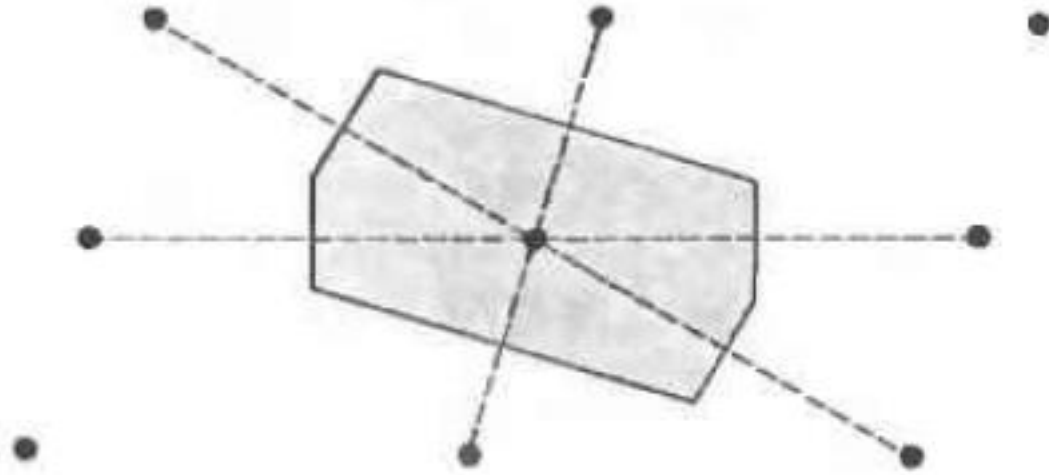
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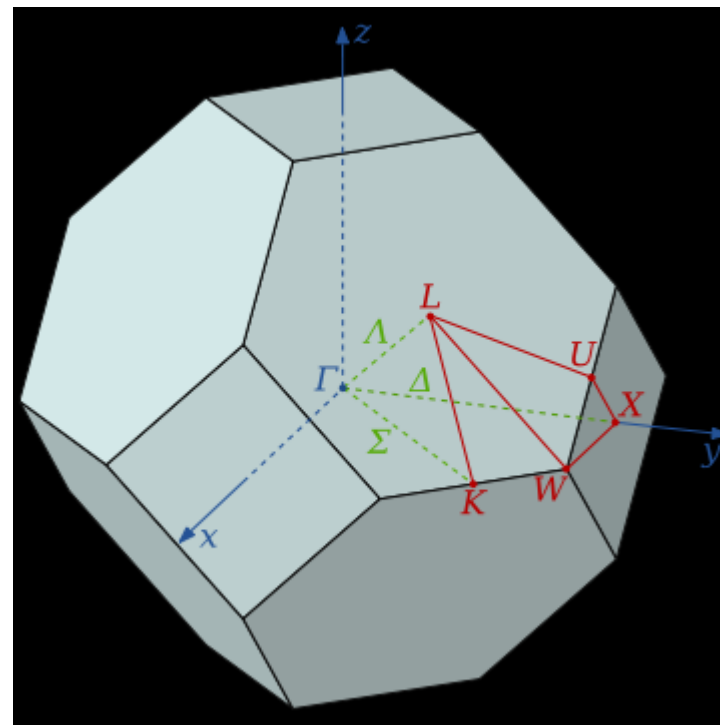
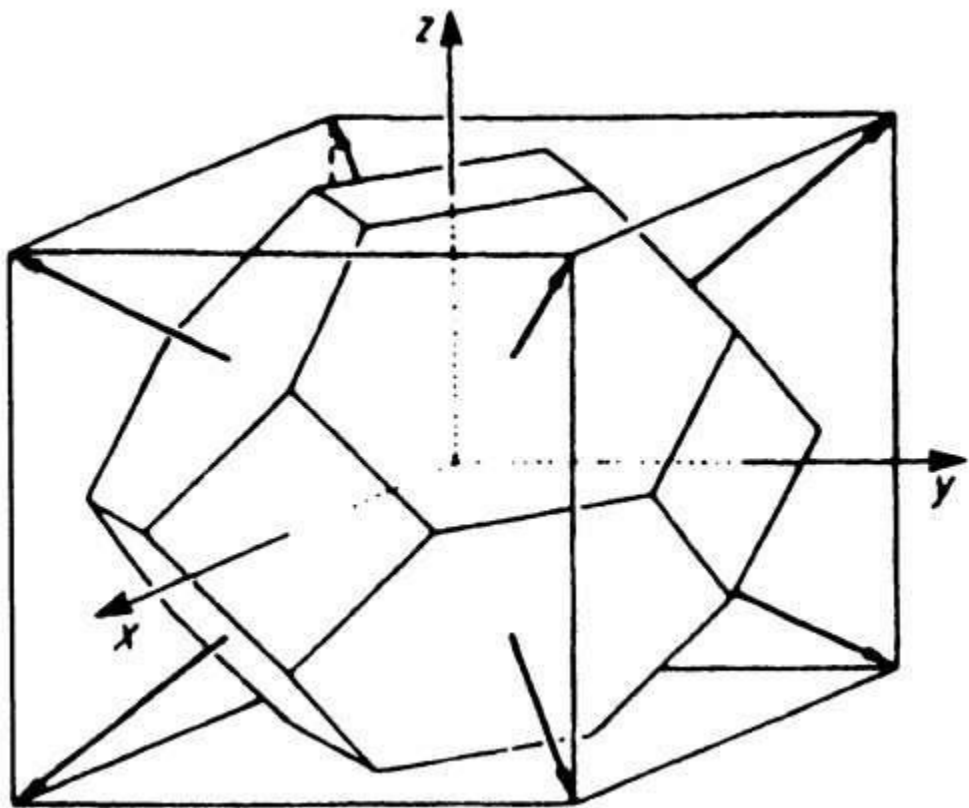
	BRAVAIS LATTICE (BASIS OF SPHERICAL SYMMETRY)	CRYSTAL STRUCTURE (BASIS OF ARBITRARY SYMMETRY)
Number of point groups:	7 ("the 7 crystal systems")	32 ("the 32 crystallographic point groups")
Number of space groups:	14 ("the 14 Bravais lattices")	230 ("the 230 space groups")

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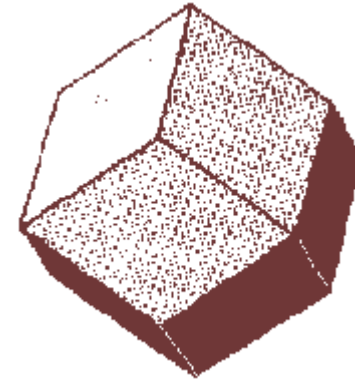
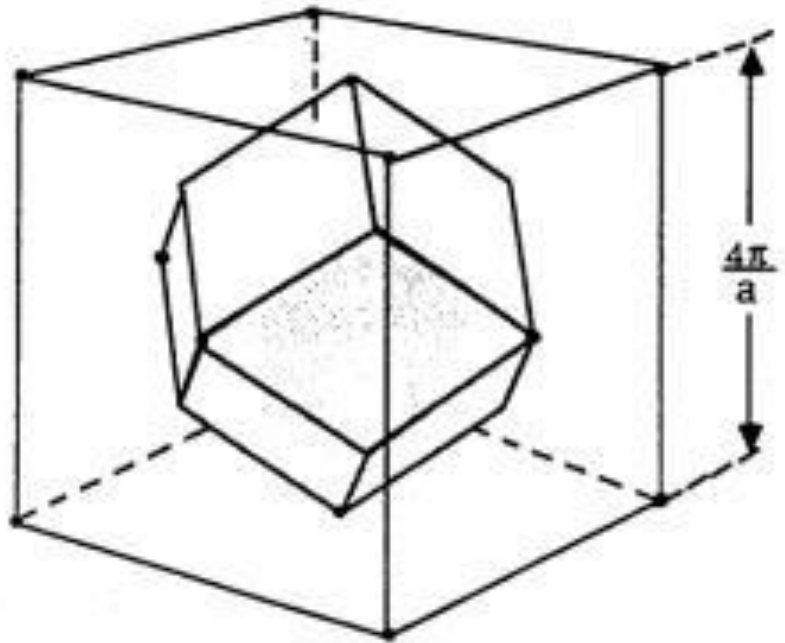
# Wigner-Seitz cell



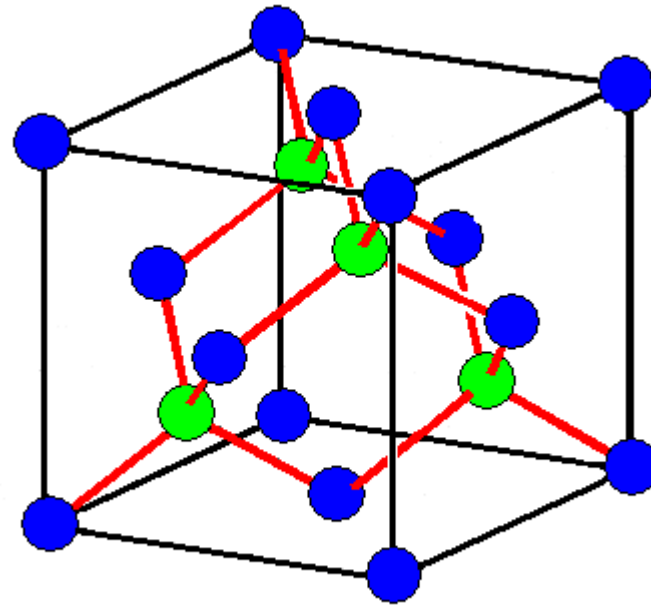
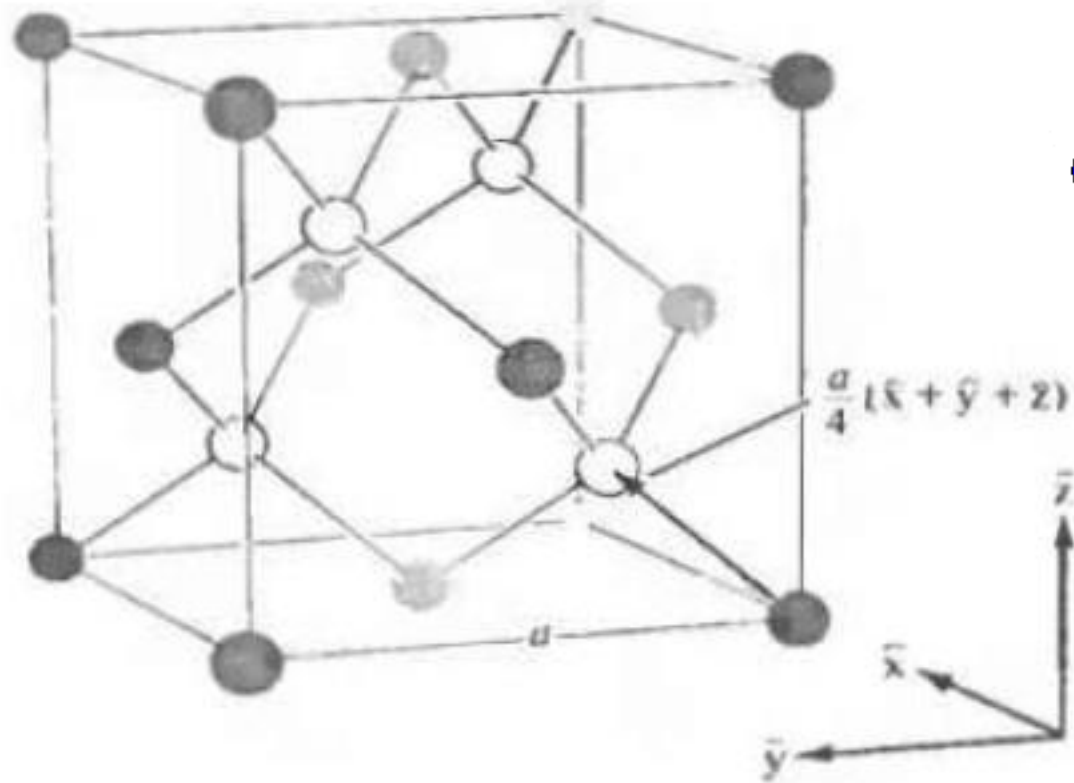
# Wigner-Seitz cell of bcc



# Wigner-Seitz cell of fcc



# Diamond

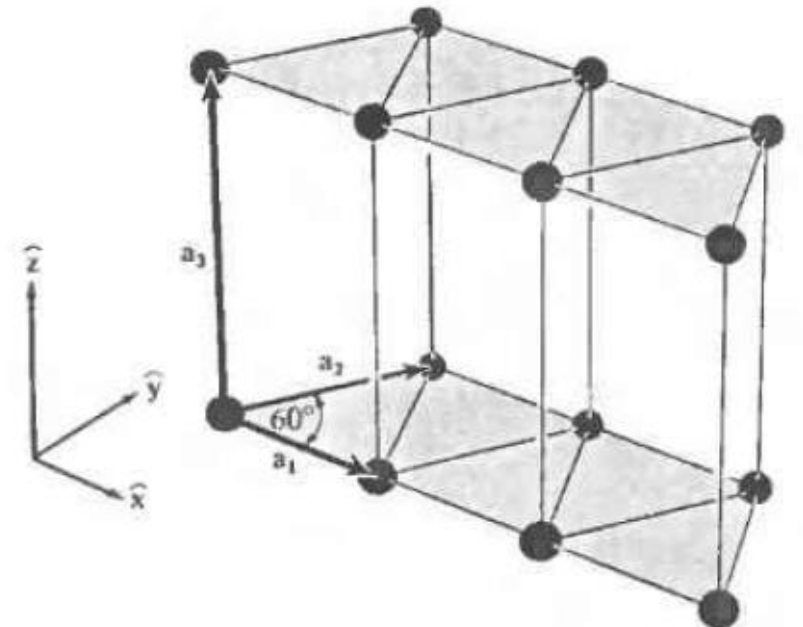
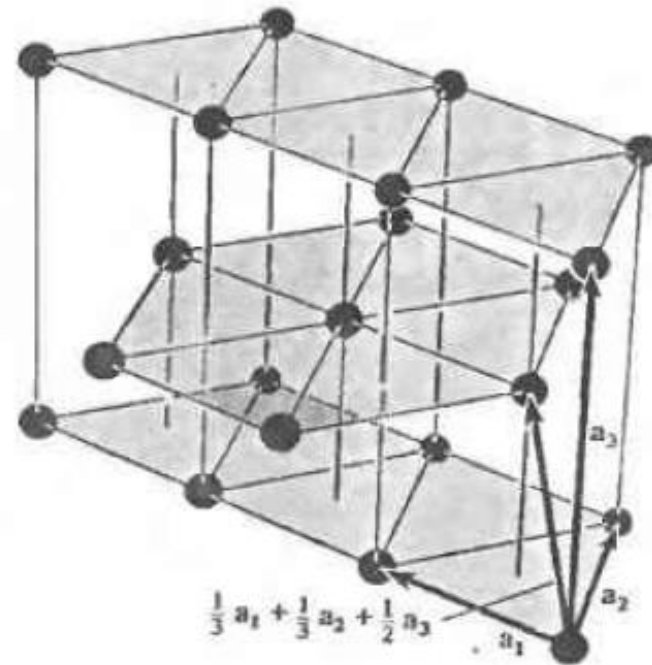


a face--centered cubic lattice with the two-point basis 0 and  $(a/4)(x + y + z)$ .

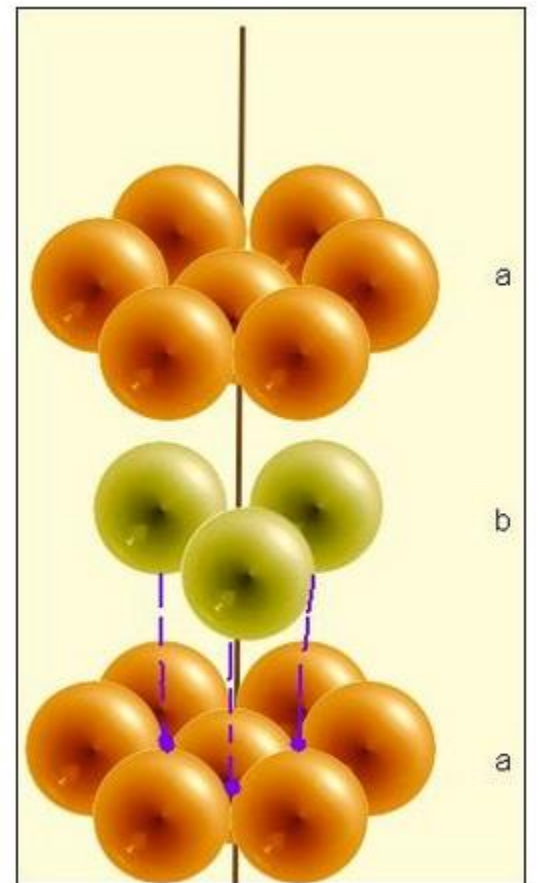
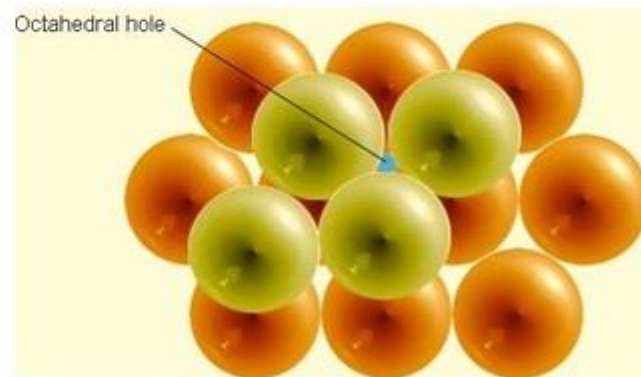
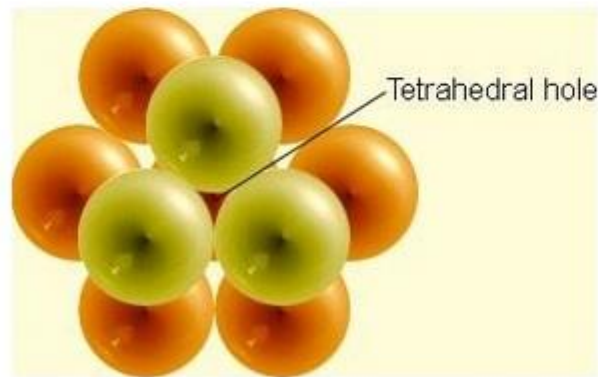
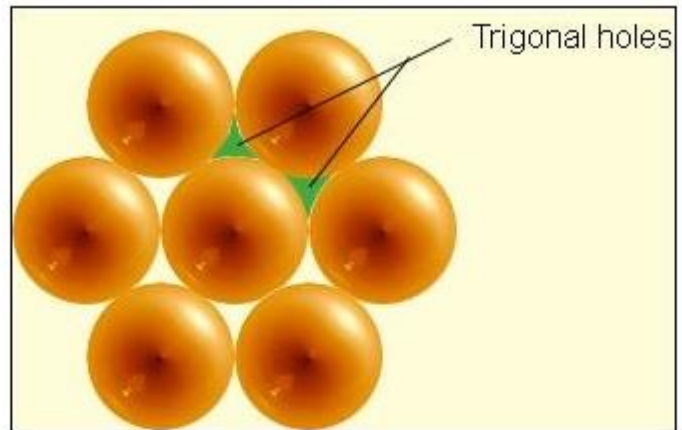
# Hexagonal Close-Packed Structure (hcp)

The hexagonal close-packed structure consists of two interpenetrating simple hexagonal Bravais lattices, displaced from one another by  $\mathbf{a}_1/3 + \mathbf{a}_2/3 + \mathbf{a}_3/2$

$$\mathbf{a}_1 = a\hat{x}, \quad \mathbf{a}_2 = \frac{a}{2}\hat{x} + \frac{\sqrt{3}a}{2}\hat{y}, \quad \mathbf{a}_3 = c\hat{z}.$$



hcp



# hcp

